illustrates the atom labeling and molecular conformation; Fig. 2 illustrates the unit-cell packing and hydrogen bonding.\*

Related literature. The title compound has shown antitumor activity (Revankar *et al.*, 1990). The structure of 6-thioguanine (Bugg & Thewalt, 1970) and 6-thioguanosine (Thewalt & Bugg, 1972) have been reported. 6-Thioguanosine has an *anti* conformation. The *syn* conformation observed in (1) is characteristic of the 8-substituted guanosines such as 7-methyl-8-oxoguanosine (Larson, Cottam & Robins, 1989), 7-methyl-8-thioxoguanosine (Larson, Henry, Kini & Robins, 1990), 8-bromoguanosine (Tavale & Sobell, 1970), 8-chloroguanosine (Birnbaum, Lassota & Shugar, 1984) and 8-methylguanosine (Hamada, Honda, Fujii, Fujiwara & Tomita, 1985). Conformational parameters follow the conventions of Altona & Sundaralingam (1972).

\* Tables of anisotropic thermal parameters, bond lengths and angles involving H atoms, torsion angles, least-squares planes and structure factor amplitudes have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 52343 (13 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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# Structure of Tetrahydrofuniculosin

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(Received 21 September 1989; accepted 24 October 1989)

Abstract.  $rel-(2'S,5'R,6'R,1''R,3''R,5''S,2'''R,3'''R,-4'''S,5'''S)-4-Hydroxy-1-methyl-3-[5'-methyl-6'-(1'',-3'',5''-trimethylheptyl)tetrahydro-2'H-pyran-2'-yl]-5-(2''',3''',4''',5'''-tetrahydroxy-1'''-cyclopentyl)-2-pyridone, C<sub>27</sub>H<sub>45</sub>NO<sub>7</sub>, <math>M_r = 495.66$ , monoclinic, P2<sub>1</sub>,

a = 19.275(2),c = 10.717 (2) Å, b = 6.968(1), $\beta = 104 \cdot 10 \ (1)^{\circ}$  $V = 1395 \cdot 9 \text{ Å}^3$ , Z = 2,  $D_r =$  $\lambda(Cu K\alpha) = 1.5418 \text{ Å},$  $\mu =$  $1.179 \text{ g cm}^{-3}$ ,  $6.474 \text{ cm}^{-1}$ , F(000) = 540, T = 298 K, final R = 0.038for 1788 unique reflections  $[F_o^2 > 1\sigma(F_o^2)]$ . The cyclopentyl ring takes an envelope conformation and the four hydroxy groups are attached on the same side of the ring. Intra- and intermolecular hydrogen bonds are observed in this moiety.

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Table 1.	Final fractional coordinates and equivalent								
isotropic	temperature factors for non-H atoms with								
e.s.d.'s in parentheses									

Table 2. Bond lengths (Å) and angles (°) with e.s.d.'s in parentheses

1·503 (6) 1·354 (5) 1·517 (6) 1.442 (4) 1.522 (7) 1.524 (5) 1.526 (8) 1.513 (6) 1.438 (6) 1.528 (5) 1.518 (8) 1.524 (7) 1.525 (6) 1.532 (9) 1.525 (8) 1.531 (6) 1.492 (9) 1.536 (9) 1.464 (7) 119.9 (3) 122.3 (3) 117.8 (3) 119.7 (3) 117-1 (3) 123.2 (4) 119.3 (4) 115.0 (3) 125.6 (3) 121.9 (3) 115-2 (3) 122.8 (4) 112.8 (3) 108.9 (3) 110-1 (3) 109-4 (4) 110.7 (4) 109·9 (4) 109.1 (4) 110-1 (5) 112·3 (5) 112·2 (5) 117.7 (4)

111-9 (5)

109.7 (4)

112.5 (6)

115.7 (5)

	e	.s.d.'s in pare	entheses	C(1)O(2)	1.427 (5)	C(16)—C(19)	
		B (1/2) 5 5	•		(1) - (3)	1.532 (6)	C(17)—O(18)
		$B_{eq} = (4/3) \sum_i \sum_j$	$\beta_{ij}\mathbf{a}_i \cdot \mathbf{a}_j$ .	$C(1) \rightarrow C(9)$	1.513 (5)	C(19)—C(20)	
	х	v	z	$B_{\rm res}$ (Å <sup>2</sup> )	C(3)-O(4)	1.421 (7)	C(19)—O(25)
C(I)	0.8561 (2)	0.674	0.1043 (4)	4.5(1)	C(3) = C(3)	1.538 (8)	C(20) - C(21)
$\hat{0}$	0.9194(2)	0.7758 (6)	0.1658 (3)	6.25 (9)	C(5) = O(6)	1.408 (5)	C(21)—C(22)
CG	0.8560 (2)	0.6521 (8)	-0.0381(4)	4.8 (1)	(1)	1.559 (6)	C(22) - C(23)
0(4)	0.8841(2)	0.8197 (6)	-0.0833(3)	6.10 (9)	C(7) = O(8)	1.414 (5)	C(22)—C(24)
as	0.8963 (2)	0.4648 (8)	-0.0490(3)	4.6 (1)	$C(\gamma) = C(9)$	1.555 (6)	C(24)—O(25)
<b>O</b> G	0.9561(2)	0.4927 (6)	-0.1008(2)	5.44 (8)	$C(9) \rightarrow C(10)$	1.514 (5)	C(24)—C(26)
cờń	0.9180 (2)	0.3775 (8)	0.0889 (4)	4.3 (1)	C(10) - C(11)	1.345 (5)	C(26)—C(27)
O(8)	0.9885 (2)	0.4236 (6)	0.1574 (3)	5.24 (8)	$C(10) \rightarrow C(17)$	1.427 (6)	C(26)—C(28)
còn	0.8599 (2)	0.4672 (8)	0.1493 (3)	4.0 (1)	V(1) = N(12)	1.369 (4)	C(28)—C(29)
C(10)	0.8702 (2)	0.4270 (7)	0.2914 (3)	3.13 (9)	N(12) - C(13)	1.400 (5)	(29) - (30)
CII	0.9203 (2)	0.5164 (7)	0.3821 (3)	3.37 (9)	C(14) = C(14)	1.382 (6)	C(29) - C(31)
N(12)	0.9308 (2)	0.4758 (5)	0.5104 (3)	3.05 (7)	C(14) = O(15)	1.254 (4)	C(31) - C(32)
C(13)	0.9874 (2)	0.5746 (7)	0.6049 (4)	3.9 (1)	C(16) = C(17)	1.420 (5)	C(32) - C(33)
C(14)	0.8910 (2)	0.3388 (7)	0.5553 (3)	3.39 (9)	$C(10) \rightarrow C(17)$	1.372 (3)	C(32) - C(34)
O(15)	0.9028 (2)	0.3073 (5)	0.6738 (2)	4.18 (7)			C(34) - C(33)
C(16)	0.8379 (2)	0.2393 (6)	0.4618 (3)	2.97 (8)	O(2) - C(1) - C(3)	107.6 (4)	C(11) $N(12)$ $C(12)$
C(17)	0.8291 (2)	0.2817 (7)	0.3339 (3)	3.34 (9)	O(2) = C(1) = C(3)	111.0 (2)	C(11) = N(12) = C(13)
O(18)	0.7806 (2)	0.1916 (5)	0.2394 (2)	4.98 (8)	C(3) - C(1) - C(9)	102.3 (3)	C(13) = N(12) = C(14)
C(19)	0.7991 (2)	0.0843 (7)	0.5150 (3)	3.38 (9)	C(1) - C(3) - O(4)	110.6 (4)	N(12) - C(14) - O(15)
C(20)	0.8454 (2)	-0.0903 (7)	0.5600 (4)	4.0 (1)	$C(1) \rightarrow C(3) \rightarrow C(5)$	106.3 (3)	N(12) - C(14) - C(15)
C(21)	0.8012 (2)	-0.2433 (7)	0.6065 (4)	4.4 (1)	O(4) - C(3) - C(5)	115.4 (4)	O(15) - C(14) - C(16)
C(22)	0.7353 (2)	- 0·2948 (7)	0.5009 (4)	4.3 (1)	$C(3) \rightarrow C(5) \rightarrow O(6)$	113.0 (4)	C(14) - C(16) - C(17)
C(23)	0.6921 (3)	<i>−</i> 0·4495 (9)	0.5490 (6)	6.7 (1)	$C(3) \rightarrow C(5) \rightarrow C(7)$	106.4 (4)	C(14) - C(16) - C(19)
C(24)	0.6925 (2)	- 0·1145 (7)	0.4569 (4)	4·0 (1)	O(6) - C(5) - C(7)	111.8 (3)	C(17) - C(16) - C(19)
O(25)	0.7362 (1)	0.0298 (4)	0.4178 (2)	3.61 (6)	C(5) - C(7) - O(8)	113.9 (4)	C(10) - C(17) - C(16)
C(26)	0.6258 (2)	-0·1387 (8)	0.3469 (4)	4.5 (1)	C(5) - C(7) - C(9)	100.6 (3)	C(10) - C(17) - O(18)
C(27)	0.6437 (3)	-0.202 (1)	0.2227 (5)	6.7 (2)	O(8)-C(7)-C(9)	113-1 (3)	C(16) - C(17) - O(18)
C(28)	0.5806 (3)	0.0432 (9)	0.3278 (5)	5.9 (1)	C(1) - C(9) - C(7)	103.4 (4)	C(16) - C(19) - C(20)
C(29)	0.5086 (3)	0.0353 (9)	0.2294 (5)	6-1 (1)	C(1) - C(9) - C(10)	118.7 (4)	C(16) - C(19) - O(25)
C(30)	0.4/06 (3)	0.229 (1)	0.2230 (7)	8.6 (2)	C(7)-C(9)-C(10)	114.4 (3)	C(20) - C(19) - O(25)
C(31)	0.4621 (2)	-0.1299 (9)	0.2550 (5)	5.5 (1)	C(9) - C(10) - C(11)	122.5 (4)	C(19) - C(20) - C(21)
C(32)	0.3897(3)	-0.1606 (9)	0.1595 (5)	5.7 (1)	C(9)-C(10)-C(17)	120.3 (3)	C(20) - C(21) - C(22)
(33) (24)	0.39/4(3)	-0.186 (2)	0.0253 (6)	9.8 (2)	C(11) - C(10) - C(17)	117.1 (3)	C(21) - C(22) - C(23)
C(25)	0.3303(3)	-0.329(1)	0.2042 (5)	6·4 (1)	C(10) - C(11) - N(12)	122-3 (4)	C(21)-C(22)-C(24)
C(33)	0.2770 (3)	-0.362 (1)	0.1291 (7)	9.5 (2)	C(23)-C(22)-C(24)	113.0 (4)	C(28)-C(29)-C(30)
					C(22)-C(24)-O(25)	111.0 (3)	C(28) - C(29) - C(31)
					C(22)-C(24)-C(26)	116-2 (4)	C(30)-C(29)-C(31)
					O(25)—C(24)—C(26)	107-1 (3)	C(29)-C(31)-C(32)
Exneri	imental Th	e title comp	ound (1) is	C(19)-O(25)-C(24)	115.0 (3)	C(31)-C(32)-C(33)	
- April	· · · ·	e due comp		C(24)-C(26)-C(27)	112.4 (4)	C(31)-C(32)-C(34)	
cataly	lic hydrog	genation pr	oducts of	C(24)—C(26)—C(28)	111.0 (4)	C(33)—C(32)—C(34)	

C(24)—C(26)—C(28)C(27)—C(26)—C(28)

C(26)-C(28)-C(29)

111.6 (4)

117.2 (5)

Experimental. The title compound (1) is one of the catalytic hydrogenation products of antibiotic funiculosin (2) with  $H_2$  and Pd--C in methanol at atmospheric pressure and ambient temperature. Colorless thin plates of tetrahydrofuniculosin were grown from acetone solution (m.p. 454-455 K, found: C 65.65; H 9.17; N 2.89%. C<sub>27</sub>H<sub>45</sub>O<sub>7</sub>N requires C 65·42; H 9·15; N 2·83%). Crystal size 0·25  $\times 0.25 \times 0.05$  mm, Enraf-Nonius CAD-4  $\kappa$ -cradle diffractometer, Cu Ka radiation, graphite mono-







C(32) - C(34) - C(35)

Fig. 1. A perspective view of the molecule with numbering scheme.

chromator,  $\theta - 2\theta$  scan with scan speed 0.47- $8.24^{\circ} \min^{-1}$  in  $\theta$ , scan width  $(0.50 + 0.14 \tan \theta)^{\circ}$ . Range of indices,  $-22 \le h \le 22$ ,  $0 \le k \le 8$ ,  $0 \le l \le 12$  $(2\theta < 130^\circ)$ . Lattice constants determined based on 18 2 $\theta$  values (13 <  $\theta$  < 34°). Variation of standard



Fig. 2. The crystal structure projected along the *b* axis. Hydrogen bonds are shown as narrow lines.

 $\sim < 0.7\%$ ; 2580 reflections measured; 1788 observed reflections with  $F_o^2 > \sigma(F_o^2)$ . Systematic absences 0k0, k odd. No corrections for absorption. Structure solved by direct methods with MULTAN (Main, Woolfson & Germain, 1971). Refined by full-matrix least squares. The locations of 45 H atoms were found on difference Fourier maps. Non-H atoms refined with anisotropic thermal parameters, H atoms with fixed isotropic thermal parameters (B =5.0 Å<sup>2</sup>).  $\sum w(|F_o| - |F_c|)^2$  minimized; w = 1.0 for  $|F_o| < 81.84, w = (81.84/F_o)^2$  for  $|F_o| \ge 81.84$ . Final R = 0.038, wR = 0.036, S = 2.40 for 497 variables, secondary-extinction factor  $g = 1.89 (6) \times 10^{-6} [|F_o|]$ =  $|F_c|/(1 + gIc)]$ ;  $\Delta/\sigma < 0.23$ , largest peak in final  $\Delta F$  map + 0.14 e Å<sup>-3</sup>; atomic scattering factors from International Tables for X-ray Crystallography (1974); programs: Enraf-Nonius SDP (Frenz, 1984), ORTEPII (Johnson, 1976). The structure of the title compound is shown in Fig. 1, crystal packing in Fig. 2. Positional parameters and equivalent values of the anisotropic temperature factors are given in Table 1, bond distances and angles are listed in Table 2.\*

**Related literature.** Funiculosin is an antiviral and antifungal antibiotic (Ando, Suzuki, Saeki, Tamura & Arima, 1969), and its structure elucidation and antifungal activity have been reported (Ando, Matsuura, Nawata, Endo, Sasaki, Okutomi, Saeki & Tamura, 1978).

\* Lists of anisotropic thermal parameters, H-atom coordinates, torsion angles, least-squares planes and structure factors have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 52364 (15 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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## 5-Hydroxyflavone\*

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(Received 16 March 1989; accepted 6 October 1989)

**Abstract.**  $C_{15}H_{10}O_3$ ,  $M_r = 238.24$ , monoclinic,  $P2_1/c$ , a = 4.746 (1), b = 18.758 (3), c = 12.944 (2) Å,  $\beta = 95.38$  (2)°, V = 1147.2 (5) Å<sup>3</sup>, Z = 4,  $D_x = 1.384$  g cm<sup>-3</sup>, Cu K $\alpha$  radiation,  $\lambda = 1.5418$  Å,  $\mu = 8.0$  cm<sup>-1</sup>, F(000) = 496, T = 293 K. Final R = 0.054 for 1003 observed reflections. The 5-hydroxy group forms a cyclic intramolecular hydrogen bond with

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the carbonyl group. The heterocyclic ring is coplanar with the benzene ring. The phenyl ring is slightly twisted, and makes a dihedral angle of  $5.7 (7)^{\circ}$  with the  $\gamma$ -benzopyrone portion of the molecule. The torsion angle for C(2')—C(1')—C(2)—O(1) is  $-5.8 (4)^{\circ}$ .

**Experimental.** The title compound (Fig. 1) was purchased from the Indofine Chemical Company. © 1990 International Union of Crystallography

<sup>\*</sup> Flavone is 2-phenyl-4*H*-benzopyran-4-one.